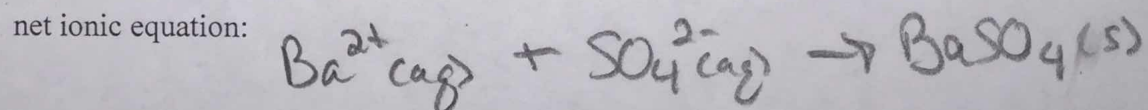
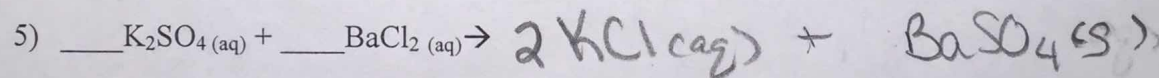
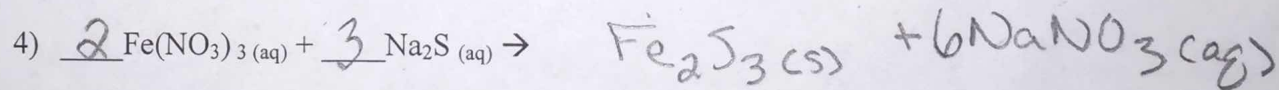
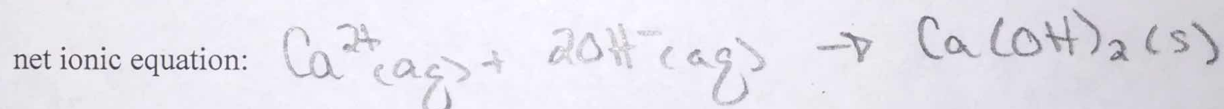
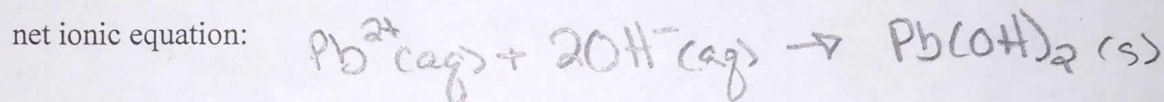
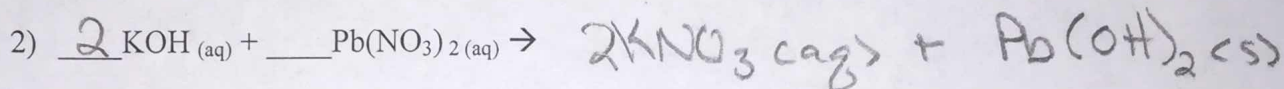
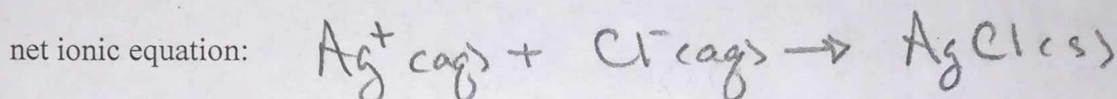
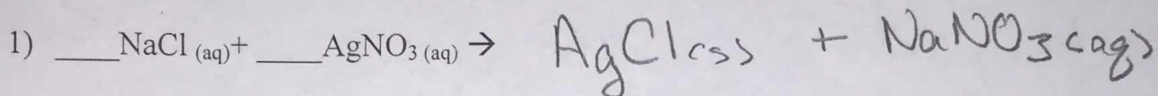
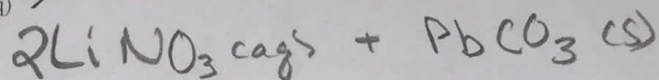
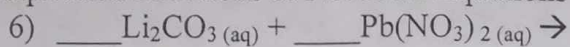


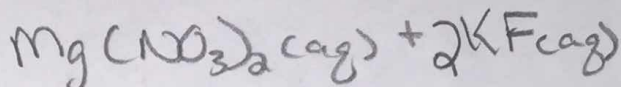
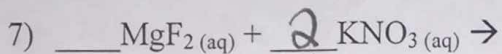
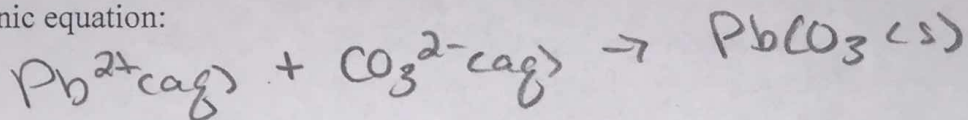
For each reaction,

- Predict the possible products for each "double replacement" reaction.
- Use the solubility rules to predict the phase of each product. Write the phase for each product.
- Balance the complete chemical equation.
- Write the net ionic equation below the complete chemical equation.



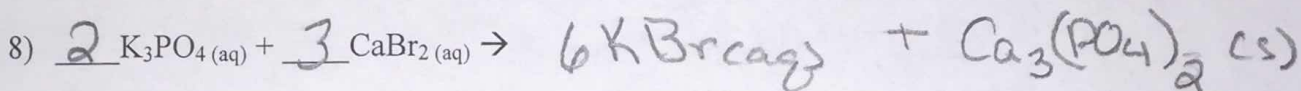


net ionic equation:

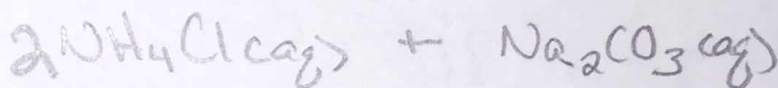
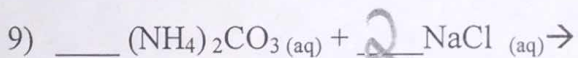
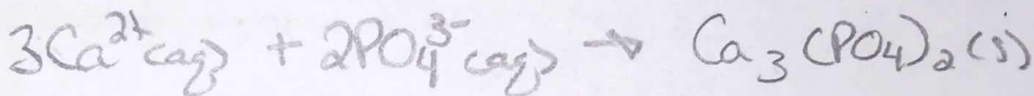


net ionic equation:

NR

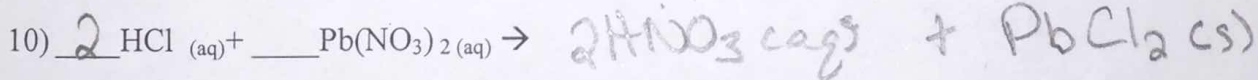


net ionic equation:



net ionic equation:

NR



net ionic equation:

